

#### Manufacturer of Specialty Chemicals

### Safety Data Sheet

Section 1. Identification

Product Identifier	Strike	Version: 6 Effective Date: 8 March, 2016
Other Means Of Identification	Drain opener	Effective Date: 8 March, 2010
Initial Supplier Identifier	Chemfax Products Ltd. 11444 – 42 Street SE Calgary, AB T2C 5C4 Tel: 403-287-2055	
Recommended Use and Restrictions On Use	Drain opener. No restrictions.	
Product Family 24 Hour Emergency	Blend Canutec (613) 996-6666	

Section 2. Hazard Identification		
Hazard Classification		
Physical Hazards	Corrosive to Metals – Category 1	
Health Hazards	Acute Toxicity (Oral) - Category 5 Skin Corrosion/Irritation - Category 1A Eye Damage/Irritation - Category 1	
Signal Word	Danger	
Hazard Statement	May be corrosive to metals. May be harmful if swallowed. Causes severe skin burns and serious eye damage.	
Precautionary Prevention	Keep only in original packaging.	
Statement	Do not breathe mists. Wash hands thoroughly after handling.	
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Statement		
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· ·	Keep only in original packaging.	



	IF INHALED: Remove person to fresh air and keep comfortable	
	for breathing. Immediately call a doctor.	
	Specific Treatment: do not induce vomiting unless directed by	
	medical personnel.	
	IF IN EYES: Rinse cautiously with water for several minutes.	
	Remove contact lenses, if present and easy to do. Continue rinsing.	
	Immediately call a doctor.	
Precautionary Storage	Store in a corrosion resistant container with a resistant inner liner.	
Statement	Store locked up.	
Precautionary Disposal	Dispose of contents / container in accordance with local	
Statement	regulations.	
Other Hazards	None	

Section 3. Composition / Information on Ingredients			
Chemical Name	Common Name or Synonyms	CAS NO. and Other Unique Identifiers	% by weight
Sulphuric AcidOil of vitriol7664-93-990 - 100			
Balance of ingredients are considered non hazardous and constitute a proprietary blend			

Section 4. First-Aid Meas	ures
Eye Contact	Immediately flush eyes with water for 30 minutes, preferably 60 minutes. Hold eyelids open during flushing. If irritation persists, repeat flushing. Do not transport victim until flushing period is complete, unless flushing can be continued during transport. Seek immediate medical attention
Skin Contact	Prompt removal of the material from the skin is essential. Remove all contaminated clothing and wash exposed areas with copious amounts of water for a minimum of 30 minutes and up to 60 minutes. Obtain immediate medical attention
Inhalation	Remove victim to fresh air. If there is difficulty breathing, seek immediate medical attention. CPR and oxygen should only be administered by trained persons.
Ingestion	Do NOT induce vomiting. Lay victim on left side to prevent aspiration of any vomit. Seek immediate medical attention. If conscious wash mouth out with water
Most Important Symptoms and Effects Both Acute and Delayed	Causes burns. Causes severe respiratory irritation if inhaled. Symptoms may include: Burning of nose and throat, constriction of airway, difficulty breathing, shortness of breath, bronchial spasms, chest pain, and pink frothy sputum. Contact may cause immediate severe irritation



	progressing quickly to chemical burns. May cause pulmonary edema.
	Symptoms may be delayed.
	Can cause blindness. May cause burns or irritation of the linings of the
	mouth, throat, and gastrointestinal tract. Swallowing a small quantity of
	this material will result in serious health hazard.
	Repeated or prolonged inhalation may damage lungs. Prolonged and
	repeated contact will eventually cause permanent tissue damage and
	effects such as erosion of teeth, lesions on the skin, tracheo-bronchitis,
	mouth inflammation, conjunctivitis, and gastritis.
Immediate Medical	Do not attempt to neutralize the acid with a weak base as the
Attention and Special	exothermic reaction may extend the corrosive injury. Do not use
Treatment	buffering agents (antacids) as they can produce significant exothermic
	reaction without significantly altering the pH.

Section 5. Fire-Fighting Mo	easures
Suitable and Unsuitable	Use extinguishing media suitable for the surrounding fire.
Extinguishing Media	Do not get water inside containers. Do not apply water stream directly
	at source of leak. Do not use a heavy water stream. A direct water
	stream will cause violent splattering and generation of heat.
Hazardous	Thermal combustion products are toxic and may include oxides of
<b>Combustion Products</b>	sulphur and irritating gases.
Specific Hazards Arising	Not flammable. Under conditions of fire this material may produce:
From the Product	Oxides of sulfur. Reacts with metals with eliberation of hydrogen.
Special Protective	Fire-fighters should wear self contained breathing apparatus and full
Equipment and	protective clothing. Use water spray to cool containers and structures
<b>Precautions For Fire-</b>	exposed to fire. Avoid direct contact of this product with water as this
Fighters	can cause a violent exothermic reaction. Closed containers exposed to
	heat may explode. Reacts with most metals to produce hydrogen gas
	which could make an explosive mixture with air.

Section 6. Accidental Release Measures		
Personal Precautions,	Chemical resistant (rubber / neoprene) gloves, coveralls and footwear.	
Protective Equipment and	Stop leak if safe to do so. Eliminate ignition sources. Evacuate	
<b>Emergency Procedures</b>	unnecessary personnel. Ventilate area. Keep upwind.	
Environmental	Do not allow spilt material to enter surface drains and watercourses.	
Precautions		
Methods and Materials	Isolate spill and stop leak. Restrict area to required and protected	
For Containment and	persons only. Ventilate area. Neutralize with lime slurry, limestone or	
Clean-Up	soda ash. Flush area with water to remove residues.	



Section 7. Handling and Storage		
<b>Precautions For Safe</b>	Handle with care, corrosive material. Empty containers may contain	
Handling	hazardous residues. Never add water to this material. Do not mix with	
	materials such as Bleach.	
<b>Conditions For Safe</b>	Store in a cool, dry, well ventilated area. Avoid direct sunlight. Keep	
Storage	containers closed when not in use. Drums may require venting to	
	release internal pressure.	

Section 8. Exposure Contro	ols / Personal Pro	otection		
Control Parameters	TWA: 8 Hr	STEL: 15 min	Ceiling	IDLH *
Sulphuric Acid	$1 \text{ mg/m}^3$			$15 \text{ mg/m}^3$
	OSHA (PEL)			(NIOSH)
	* Immediately I	Dangerous to Life and	Health	
Exposure Controls	Local exhaust v	entilation		
Appropriate Engineering	Provide sufficient ventilation to keep vapors below the permissible			
Controls	exposure limit. Ensure adequate ventilation, especially in confined			
	areas. Packaging and unloading areas and open processing equipment			
	may require me	chanical exhaust system	ms. Corrosion-	proof
	construction rec	ommended.		
Individual Protective				
Measures				
Eye / Face Protection	Safety glasses			
Skin Protection	Chemical resistant (rubber/ neoprene) gloves, coveralls and footwear			
<b>Respiratory Protection</b>	Air purifying respirator fitted with cartridges for acid vapours and			
	mists.			

Section 9. Physical and Chemical Properties		
Appearance	Dark brown liquid	
Odour	Pungent odour	
Odour Threshold	Not available.	
pH	0.3 (1 N aqueous solution)	
Flash Point	>100 °C	
<b>Boiling Point and Boiling Range</b>	150 - 330 °C	
Melting Point and Freezing Point	-40 to -1.1 °C	
Evaporation Rate	Not determined	
Flammability (solid, gas)	Not applicable	
Upper and Lower Flammability or	No data	
Explosive Limits		



Vapour Pressure	0.002 to 1.2 @ 20 °C
Vapour Density	3.4
Relative Density	1.775
Solubility	Soluble
Partition co-efficient, n-	No data
Octanol/Water	
Auto-ignition Temperature	No data
<b>Decomposition Temperature</b>	340 °C
Viscosity	No data

Section 10. Stability and Reactivity	
Reactivity	Extremely reactive with metals, alkalis, reducing agents and many other organic and inorganic chemicals. Hazardous gases such as hydrogen cyanide, hydrogen sulfide and acetylene are evolved on contact with chemicals such as cyanides, sulfides and carbides respectively. Contact with combustible organic matter may cause fire or explosion. Dilution with water generates excessive heat and spattering or boiling may occur. Always add acid to water, NEVER ADD WATER TO ACID. Corrosive to most metals including mild steel, copper, aluminum, zinc, etc., especially when diluted to below 90%.
Chemical Stability Possibility of Hazardous	Will not occur
Reactions	win not occur
Conditions to Avoid	Excessive temperatures. Avoid contact with water. Do not store in humid places. Material is hygroscopic (readily absorbs water from the atmosphere).
Incompatible Materials	Violently reactive with: sodium chlorite, reducing agents, strong bases, combustibles, metals, alkali metals and their hydrides, organic materials, aluminum and its alloys, copper and its alloys, cast iron, mild steel, titanium. Material will attack some rubber, plastics and coatings.
Hazardous Decomposition Products	Not expected to decompose. Material will react with metals as listed above and produce hydrogen gas.

Section 11. Toxicological Information			
Component Toxicity	LD50 Oral	LD50 Dermal	LC50 Inhalation
Sulphuric Acid	2.14g/kg (Rat)		255mg/m <sup>3</sup> (Rat), 4h
Likely Routes of Exposure	2		



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<u> </u>
May cause severe skin burns.
May cause severe burns and even permanent blindness.
May be corrosive to the respiratory passage. Vapours may cause
pulmonary oedema (fluid in the lungs). Symptoms can be delayed
for several hours.
This product causes severe burning and pain in the mouth, throat
and abdomen. Vomiting, diarrhea and perforation of the esophagus
and stomach lining may occur. Prolonged and repeated exposure
may cause discolouration and erosion of the teeth.
Oral LD50 Based on ATE data, the classification criteria are not
met. ATE > 2000 mg/kg.
Dermal LD50 Based on ATE data, the classification criteria are not
met. ATE > 2000 mg/kg.
Vapor LC50 Based on ATE data, the classification criteria are not
met. ATE > 20 mg/l.
Respiratory system
Not classified
Not classified
Not classified
Not classified
Not classified
This substance has no evidence of carcinogenic properties.
Not classified
None known
None known



#### Section 12. Ecological Information

Harmful to aquatic life at low concentrations and is primarily	
associated with low pH. 24 hr TLm = $24.5$ mg/L (Bluegill) 48 hr	
TLm = 49 mg/L (Bluegill) 48 hr LC50: 100 – 300 mg/L (Flounder)	
Will not persist	
Will not bioaccumulate	
Not available	
Not available	
None known	

Section 13. Disposal Considerations	
<b>Disposal Considerations</b>	Dispose of contents / container in accordance with local regulations.

Section 14. Transport Information	
UN Number	UN1830
UN Proper Shipping Name	Sulphuric acid
Transport Hazard	8
Class(es)	
Packaging Group	П
Environmental Hazards	Not applicable
Bulk Transport	Not applicable
Special Precaution	Not applicable
DOT Erg#	137

Section 15. Regulatory Information	
Canada – DSL Inventory	All components of this product are either on the Domestic Substances
	List (DSL) or Non-Domestic Substances List (NDSL) or exempt
TSCA	All components of this product are either on the Toxic Substances
	Control Act (TSCA) Inventory List or exempt
<b>Additional Information</b>	None



#### **Section 16. Other Information**

NFPA Rating	Health-2/ Flammability-0/Reactivity-2/Special Hazard-Not applicable
HMIS Rating	Health-2/Flammability-0/Reactivity-2/Personal Protection-See Section 8.
Prepared by:	Chemfax Products Ltd., Technical Department
<b>Date Prepared:</b>	5 January, 2012
Date of Latest Revision:8 March, 2016	

#### Disclaimer

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